

# ORGANOMETALLIC COMPLEXES BEARING SALBEN AND SALMEN LIGANDS AND THEIR APPLICATIONS IN CATALYSIS

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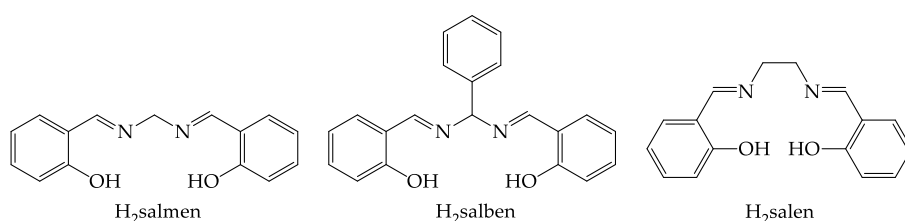
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## ABSTRACT

H<sub>2</sub>salmen (*N,N'*-bis(salicylidene)methanediamine) and H<sub>2</sub>salben (*N,N'*-bis(salicylidene)phenylmethanediamine) (**Figure 1**) are Schiff bases obtained from condensation between salicylaldehydes (salH) and methanediamine or phenylmethanediamine, and can be used as N<sub>2</sub>O<sub>2</sub> tetradentate ligands towards the complexation of metal ions like copper [1], cobalt [2,3], iron and manganese [4].



**Figure 1.** Molecular structure of H<sub>2</sub>salmen, H<sub>2</sub>salben and H<sub>2</sub>salen.

Differently from the most famous tetradentate Schiff base H<sub>2</sub>salen (**Figure 1**), derived from salH and ethylenediamine (en), which possesses the same tetradentate N<sub>2</sub>O<sub>2</sub> donor atom set, H<sub>2</sub>salmen and H<sub>2</sub>salben have one carbon

atom bridging the two imine groups. This is crucial in the formation of di- or polynuclear metal complexes [1-4], and in the case of copper(II) [1] and cobalt(II) [2], the metal ion coordination is between tetrahedral and square planar (the latter, typical for mononuclear salen complexes). Those properties can be exploited for obtaining efficient catalysts, thanks to the flexibility of the coordination environment and the possibility to obtain mixed oxidation state species [2]. My PhD project will be then devoted to the study of metal-salmen/salben complexes for potential application in catalysis, in analogy to the well-known catalytic properties of metal-salen derivatives [5]. First, efforts will be directed toward the synthesis of H<sub>2</sub>salmen and H<sub>2</sub>salben ligands with modulated steric and electronic features with the help of chemometrics methods, followed by the coordination of first row transition metals. The catalytic properties of the new complexes will be tested on benchmark or redox reactions as documented for salen-M complexes (M = Co, Cu, Fe, Mn). Chemometrics tools will be a valid support also at this stage to rationalize the findings and model the catalytic activities of the systems. The reaction mechanisms of the tested catalysts will be also explored to help improving the efficiency upon modulation of the reaction conditions.

## REFERENCES

- [1] Alessandro Pasini et al., *J. Chem. Soc., Dalton Trans.*, **2000**, 3467–3472.
- [2] Brunetto Chiari et al., *J. Chem. Soc., Dalton Trans.*, **2001**, 3611–3616.
- [3] Brunetto Chiari et al., *J. Chem. Soc., Dalton Trans.*, **2002**, 4672–4677.
- [4] Luca Rigamonti et al., *Int. J. Mol. Sci.*, **2020**, *21*, 7882.
- [5] Pier Giorgio Cozzi, *Chem. Soc. Rev.*, **2004**, *33*, 410–421.
- [6] G. Coates et al., *Angew. Chem. Int. Ed.*, **2003**, *42*, 5484–5487.
- [7] T. Katsuki et al., *ChemInform*, **2003**, *03*, 281–297.
- [8] T. Nguyen et al., *Organometallics*, **2003**, *22*, 3374–3381.
- [9] I. Maleev et al., *Tetrahedron*, **2001**, *57*, 2491.